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the establishment  
of a criterion for  
determination of  
the feasibility or  
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transformation. In  
this manner,  
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typically used to  
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occur in the  
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formation of liquid water is  $-237 \text{ kJ / mol}$ , calculate the potential for the formation of hydrogen and oxygen from water. To solve this problem we must first calculate  $\Delta G$  for the reaction, which is  $-2(-237 \text{ kJ / mol}) = 474 \text{ kJ / mol}$ . Knowing that

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$\Delta G = -nFE^{\circ}$  and  $n = 4$ , we calculate the potential is  $-1.23$  V.

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