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Covalent Bond | Ionic Bonding | Class 11  
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Covalent Bond: C-F bond . Both are nonmetals and have a high electronegativity). Therefore, both are going to gain electrons. In order for them both to gain electrons, they must share electrons....

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Chemistry Bonding Packet Answers = a  
chemical bond resulting from

\_\_\_\_\_ between positive and negative ions a)

In a purely ionic bond, one or more  
electrons are transferred from one atom to  
another 2) \_\_\_\_\_ = a chemical bond  
resulting from the \_\_\_\_\_ of electrons

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between two atoms a) \_\_\_\_\_ = a covalent bond in which the bonding

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Chemical Bonding SECTION 1 SHORT  
ANSWER Answer the following questions  
in the space provided. 1. a A chemical bond  
between atoms results from the attraction

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between the valence electrons and of different atoms. (a) nuclei (c) isotopes (b) inner electrons (d) Lewis structures 2. b A covalent bond consists of (a) a shared electron. (c) two different ions.

## 6 Chemical Bonding

Determine if it is an ionic bond or a covalent

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bond. Show the work and the final answer.

... Chemistry – Bonding Packet

Name:\_\_\_\_\_ Hr:\_\_\_\_\_ Page 1. 1 Updated  
1/20/2014. Show work here. Your choices  
can be (and you will use some them more  
than once): Covalent Metallic Bond  
Network Solid ...

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Bond: Na-F bond. Na is a metal and has a  
low electronegativity (attraction of  
electrons) while F has a high  
electronegativity. Therefore, fluorine will  
take an electron from sodium in an...  
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Atomic Structure and Bonding. These worksheets are self-explanatory. H= Bonds Broken Bonds Formed = [H-H + Cl-

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Cl]-[2(H-Cl)] = [436. Unless you have some understanding of the true nature of the double. Please feel free to ask any questions before purchasing. Open World Key (1). nonpolar covalent compound. chemistry-a-bonding-packet-answers.

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