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# Activity Series Chemistry Lab Answers

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& Displacement Reactions  
#30 Chemistry Experiment 4.2  
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Lab Answer Single Replacement  
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Solutions of Metallic Salts  
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~~Lab Answers~~  
~~Making Predictions Using~~  
~~Reactivity Series | Reactions |~~  
~~Chemistry | FuseSchool~~ Activity  
Series Lab Activity Series  
Chemistry Lab Answers

$\text{Ni(s)} + \text{Pb(NO}_3)_2(\text{aq}) \rightarrow \text{Ni(NO}_3)_2(\text{aq}) + \text{Pb(s)}$   
 $\text{Ni(s)} + \text{Fe(NO}_3)_3(\text{aq}) \rightarrow \text{NR (no reaction)}$

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the descriptions that accompany the activity series of metals, a given metal is also capable of undergoing the reactions described below that section. For example, lithium will react with cold water, replacing hydrogen.

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7.11: The Activity Series -  
Chemistry LibreTexts  
Grade 11 Chemistry Activity  
Series of Metals Lab Problem:  
What is the order of reactivity of  
the metals copper, iron,  
magnesium, and zinc in single  
displacement reactions? Materials:

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Wellplate/Spotplate Small pieces of magnesium, iron, zinc and copper metal Dilute solutions of hydrochloric acid, copper (II) sulfate, zinc chloride, magnesium chloride, iron (III) sulfate Wash bottle with ...

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Activity Series Lab (akey) - Grade  
11 Chemistry Activity ...

Part 1. An Activity Series for  
Some Metals  $\text{Cu}^{2+}$  (aq)  $\text{Mg}^{2+}$   
(aq)  $\text{Pb}^{2+}$  (aq)  $\text{Zn}^{2+}$  (aq)  $\text{Ag}^{+}$   
(aq)  $\text{Cu}$  (s) × No reaction No  
reaction No reaction -  $\text{Cu}$  is  
oxidized -  $\text{Ag}$  is reduced  $\text{Mg}$  (s) -

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Color of Mg metal fades away -  
Bubbles forming - Cu is reduced -  
Mg is oxidized  $\times$  - Bubbles form  
during the reaction - Pb is  
extracted - Pb is reduced

An Activity Series - Judy Chen  
When an atom gains electrons, it is

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reduced. Metals higher on the activity series are more likely to react relative to those lower on the activity series. The activity series can be used to predict products of reactions, and to predict if a reaction will even occur. In this experiment, different



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metals were tested for their reactivity. It was recorded if a reaction occurred or not, so that an activity series could be created.

Data & Results

Chemistry Lab Report (The Activity Series) – Sarah Jackson

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<http://socratic.org/questions/what-are-metal-activity-series>. We can use the series to predict whether a metal displacement reaction will occur. For example, zinc is above copper in the series. We predict that placing a strip of zinc metal in a copper(II) sulfate solution will

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Lab Answers produce metallic copper and zinc sulfate. Copper is below zinc in the series.

Metal Activity Series - Chemistry  
| Socratic

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# Where To Download Activity Series Chemistry

Activity Series of Metals & Elements - Chemistry Students can add their own graphs here. Activity Series The purpose of this lab was to observe the reactions of metals and the importance of the activity series. Magnesium-HCL Zinc-HCL Iron-

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HCL Tin-HCL Copper-HCL Copper-  
AgNO<sub>3</sub> Grade and

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Read Online Activity Series  
Chemistry Lab Answers reactive.

It was known before the  
experiment that the metals used in  
the experiment are placed in the  
activity series from most active to  
least active as follows: magnesium,

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Lab Answers, aluminum, zinc, and copper.

Activity Series Lab Answers |  
SchoolWorkHelper List the metals  
in order of their

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When the student activity sheet  
/tutorial is used with computer  
simulation and the computer

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animations representing reactions at the particle level (atom level), and when students have the opportunity to do an activity series of metal experiment in the laboratory it is an effective way of exposing students to all three levels of representation in Alex

# Where To Download Activity Series Chemistry Lab Answers Johnstone's triangle: microscopic ...

Activity Series of Metals  
Computer Simulation |  
Chemdemos  
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Activity Series Chemistry Lab  
Answers - [go.rotorxracing.com](http://go.rotorxracing.com)  
1 Malleability is ability to be  
pounded flat without shattering. 2  
Ductility is the ability to be drawn  
out into a fine wire. 3 Valence

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Lab Answers  
electrons are those in the outermost occupied ' shell ' in an atom ' s electron configuration. The Activity Series of Metals Page 2 of 13.

Lecture Notes 7 + Experiment 7 :  
ACTIVITY SERIES OF METALS

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# Where To Download Activity Series Chemistry Lab Answers home

home [intro.chem.okstate.edu]

The activity of 6 metals, Sn, Fe, Mg, Cu, Zn, and Ca were tested using 3M HCl and water separately. Observations were



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used to qualitatively rank the  
metals ...

Activity Series Of Metals  
Complete Lab - YouTube  
CHEMISTRY SINGLE  
REPLACEMENT REACTION  
WORKSHEET Using the Activity

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Series Table, complete the following reactions by writing the products that are formed. Be sure to Balance each equation. If No single replacement reaction occurs, write NR to the right of the arrow.

1.  $\text{Ag} + \text{KNO}_3$  2.  $\text{Zn} + \text{AgNO}_3$  3.  $\text{Al} + \text{H}_2\text{SO}_4$  4.  $\text{Cl}_2 +$

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K15. Li + H<sub>2</sub>

## CHEMISTRY SINGLE REPLACEMENT REACTION WORKSHEET

This chemistry video tutorial explains the activity series of metals and elements such as

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hydrogen. It shows you how to tell  
if a single replacement reactio...

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d979604cf528ddb0d