

# File Type PDF Acids Bases And Salts Answers Key

## Acids Bases And Salts Answers Key

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Acids Bases and Salts Quiz **ACIDS**  
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9th std Science 5.Acids, Bases and  
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Chapter 5: Acids, Bases and Salts -  
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**answers** *Acids, Bases and Salts* |  
*Ch-2, Part-3 | Class 10 ncert science |*  
*explained in hindi* **Acid , Base and**  
**Salt ?? ??? 39 ?????? || Lucent**  
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**acid, base, and salt || ???? , ?????? ,**

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Bases and Salts | Class 7 Science

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Bases And Salts Answers

Solution: The reaction of the acid + base gives a product of salt + water, which is considered as neutralization reaction. Examples:  $\text{NaOH} + \text{HCl} \rightarrow$

$\text{NaCl} + \text{H}_2\text{O}$

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~~NCERT Solutions for class 10 chapter  
2 Acids, bases and salt~~

Class 10 Science Chapter 2 Extra  
Questions and Answers Acids Bases  
and Salts. Extra Questions for Class  
10 Science Chapter 2 Acids Bases  
and Salts with Answers Solutions.  
Acids Bases and Salts Extra  
Questions Very Short Answer Type.  
Question 1. Two solutions A and B  
have pH values of 5 and 8  
respectively. Which solution will be  
basic in nature ...

~~Acids Bases and Salts Class 10 Extra  
Questions with ...~~

Ans:- Salts are classified into four  
categories on the basis of the strength  
of acid and base. they are: Salt of  
strong acid and strong base.e.g NaCl,  
KCl; Salt of strong acid and weak

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base. e.g.  $\text{NH}_4\text{Cl}$ ,  $\text{NH}_4\text{I}$ ; Salt of weak acid and strong base. e.g.  $\text{CH}_3\text{COONa}$ ,  $\text{COO}^- \text{Na}^+$  + Salt of weak acid and weak base. e.g.  $\text{CH}_3\text{COO}^- \text{NH}_4^+$  + How is common salt found in nature?

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2. Bases react with acids to produce salts and water; water to produce acids and salts; salts to produce acids and water; neither acids, salts, nor water

## ~~Acids Bases and Salts Multiple Choice Questions and Answers~~

The salts of strong acids and weak bases give acidic solution having pH less than 7. Example,  $\text{NH}_4\text{Cl}$ , Ammonium Chloride will have pH less than 7. The salts of weak acids and strong bases give basic solution having pH more than 7. Example,  $\text{Na}_2\text{CO}_3$ , Sodium Carbonate will have pH more than 7. Question 7.

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## ~~Acids And Bases Answer Key~~ ~~Teacher Worksheets~~

Answer A reaction in which an acid and base react with each other to give a salt and water is termed as neutralization reaction. For Example: (i)  $\text{NaOH} + \text{HCl} \rightarrow \text{NaCl} + \text{H}_2\text{O}$  (ii)  $\text{HNO}_3 + \text{KOH} \rightarrow \text{KNO}_3 + \text{H}_2\text{O}$  15. Give two important uses of washing soda and baking soda. Answer Two important uses of washing soda are:



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~~NCERT Solutions for Class 10th  
Science: Ch 2 Acids, Bases ...~~

Answer. (a) HCl is acid and NaOH is base whose combination forms the common salt. Its formula is NaCl (Sodium chloride). It is obtained from sea water. (b) Rock salt is the common name for the mineral "halite". Its chemical formula is NaCl. It may be white or light blue or yellow depending upon impurities present in it.

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Important Questions ...~~

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2018 Asst Jr Engineer Exam Pattern.  
phenylamine aniline as an amine  
chemguide.

## ~~Acids Bases And Salts Note Taking Answers~~

(A) inorganic acid (B) organic acid (C)  
soft organic acid (D) strong inorganic  
acid. Answer. Answer: (a) 5. When  
milk of magnesia reacts with acetic  
acid it produces (A) basic salts (B)  
acidic salts (C) neutral salt (D)  
complex salt. Answer. Answer: (c) 6.  
Which of the following phenomena will  
occur when a small amount of acid is  
added to water ...

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## ~~MCQ Questions for Class 10 Science Acids Bases and Salts ...~~

Turmeric is a natural indicator. On adding its paste to acid and base separately, which colours would be observed. (a) Yellow in both acid and base (b) Yellow in acid and red in base (c) Pink in acid and yellow in base (d) Red in acid and blue in base. Answer. Answer: (b) Yellow in acid and red in base

## ~~MCQ Questions for Class 7 Science Chapter 5 Acids, Bases ...~~

MCQ Questions for Class 7 Science with Answers were prepared based on the latest exam pattern. We have Provided Acids, Bases and Salts Class 7 Science MCQs Questions with Answers to help students understand the concept very well. Acids, Bases

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and Salts Class 7 MCQs Questions with Answers. Choose the correct answer: Question 1.

## ~~MCQ Questions for Class 7 Science Chapter 5 Acids, Bases ...~~

Acids, Bases and Salts In this first quiz on GCSE Chemistry we take a look at one vital part of the subject - acids, alkalis, bases and salts. In it we find out the difference between alkalis and bases, how acids and alkalis react with one another, how neutralisation can produce salts and the relevance of pH values.

## ~~GCSE Acids, Bases and Salts | Revise pH Values and Alkalis~~

Acids, bases and salts An acid is a compound that is defined by its physical and chemical properties. Acids taste sour and react with metals

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and polyatomic ions called carbonates. A carbonate is a charged cluster of Carbon and Oxygen atoms.

~~Acids, bases and salts. 7th Grade  
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Chapter 2 Acids, Bases ...~~

Acid + Base  $\rightarrow$  Salt + Water.

Question 3: Rain becomes acidic due

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to. carbon dioxide. sulphur dioxide.

nitrogen dioxide. all of the above.

Correct Option is : 4. Solution : Rain becomes acidic because of carbon dioxide, sulphur dioxide and nitrogen dioxide. They dissolve in rain drops to form carbonic acid, sulphuric acid and nitric acid ...

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