

13 3 Practice Problems Chemistry Answers

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13-3 Practice Problems the factory is 761.3 mm Hg. The pressure inside the bag is the same. What is the pressure inside the bag of potato Chips in Pa? The same bag of potato chips from Problem 5 is shipped to a town in Colorado, where the atmospheric pressure is 99.82 kPa. What is the difference (in Pa) between the pressure in the bag and the atmospheric

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Chapter 13 19 Prentice-Hall, Inc. 13-3 Practice Problems 1. A gas occupies a volume of 458 mL at a pressure of 1.01 kPa and temperature of 295 K. When the pressure is changed, the volume becomes 477 mL. If there has been no change in temperature, what is the new pressure? 6. What will be the volume of a gas sample at 309 K if its volume at 215 K is 3.42 L?

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13.3: Factors Affecting Solubility Conceptual Problems Use the kinetic molecular theory of gases discussed in Chapter 10 to explain why the solubility of virtually all gases in liquids decreases with increasing temperature.

13.E: Properties of Solutions (Exercises) - Chemistry ...
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Practice Chemistry with Worked Chemistry Problems
13-4 Practice Problems 1. What volume would be occupied by 100. g of oxygen gas at a pressure of 1.50 atm and a of 250C? ma O '082/ • (IF *273k) 2. An air-filled balloon has a volume of *5 L at 0.94 and 250C. Soon after, the pressure changes to 0.99 and the temperature changes to 0CC. What is the new volume of the balloon? /.26 1.26 L ...

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